Nakadomarin A, a Novel Hexacyclic Manzamine-Related Alkaloid from Amphimedon Sponge

Jun'ichi Kobayashi,* Daisuke Watanabe, Naoko Kawasaki, and Masashi Tsuda

Faculty of Pharmaceutical Sciences, Hokkaido University, Sapporo 060, Japan

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A novel manzamine-related alkaloid consisting of unprecedented 8/5/5/15/6 ring system, nakadomarin A (1), has been isolated from an Okinawan marine sponge Amphimedon sp., and the unique structure containing a furan ring was elucidated on the basis of the spectroscopic data. The relative stereochemistry was deduced from the NOE data and proton couplings, and a plausible biogenetic path of **1** through ircinal A was proposed.

Recently a series of unique polycyclic alkaloids with intricate skeletons have been isolated from marine Haplosclerid sponges of genera Haliclona, Xestospongia, Amphimedon, and Reniera,¹ among which the representative alkaloids, manzamines A^{2,3} and B,⁴ are characterized by a penta- or tetracyclic nitrogen-containing ring system bound to a β -carboline, respectively. These unusual ring systems have attracted great interest as one of the most challenging targets for total synthesis. Our continuing search for biogenetic precursors of manzamines A-C resulted in the isolation of several novel alkaloids, ircinals A and B,5 keramaphidins B⁶ and C,7 ircinols A and B,8 and keramamine C7 from an Amphimedon sponge. Ircinals and keramaphidin B correspond to tetra- and pentacyclic biogenetic precursors of manzamines A and B, respectively, proposed by Baldwin and Whitehead.⁹ Further investigation of biogenetically related compounds to manzamines from another Amphimedon sponge led to the isolation of nakadomarin A (1), a novel furan-containing hexacyclic alkaloid consisting of an unprecedented 8/5/5/15/6 ring system. In this paper we describe the isolation and structure elucidation of 1 and propose a biogenesis of 1 through ircinal A.

The sponge Amphimedon sp. (SS-264) collected off Kerama Islands, Okinawa, was extracted with MeOH. EtOAc-soluble materials of the MeOH extract were purified by silica gel chromatographies (CHCl₃/MeOH and then cyclohexane/acetone/Et₂NH) to afford nakadomarin A (1, 1.8 \times 10⁻³ %, wet weight) as a free base, together with known manzamine alkaloids.

Nakadomarin A (1) { $[\alpha]^{25}_{D} - 16^{\circ}$ (*c* 0.12, MeOH)} was obtained as a colorless amorphous solid and the molecular formula was established as C₂₆H₃₆N₂O by HREIMS $(m/z 392.2826, M^+, \Delta - 0.2 \text{ mmu})$. The ¹H and ¹³C NMR



data (Table 1) revealed the presence of eight sp² carbons, which were attributed to two di-, one tri-, and one tetrasubstituted double bonds, and eighteen sp³ carbons containing three methines, fourteen methylenes, and one quaternary one. Since 4 out of 10 elements of unsaturation implied by the molecular formula were accounted for, 1 was inferred to possess six rings. Three partial structures **A**-**C**, one isolated methylene (C-12), and two



methines (C-3 and C-6) were assigned by detailed analyses of ${}^{1}H-{}^{1}H$ COSY. HOHAHA, and HMQC spectra. The ¹³C chemical shifts of C-6 (δ 75.9), C-10 (δ 46.6), C-12 (δ 60.6), C-14 (δ 60.4), C-20 (δ 51.7), and C-29 (δ 59.6) suggested that these carbons were adjacent to a nitrogen atom. Geometries of two disubstituted $\Delta^{15(16)}$ and $\Delta^{24(25)}$ double bonds were elucidated to be both Z-configuration by NOESY correlations for H-15/H-16, H-14/H-17a, H-24/ H-25, and H₂-23/H₂-26 as well as ¹H-¹H coupling constants ($J_{15,16} = 10.1$ Hz and $J_{24,25} = 10.8$ Hz). Connection among C-6, C-8, C-12, and C-13 via C-7 was implied by HMBC cross-peaks for H-6/C-12, H-8/C-12, H-8/C-13, H₂-12/C-6, H-13^β/C-6, H-12^β/C-7, and H₂-13/C-7. Longrange C-H couplings for H-10 β /C-12, H-10 α /C-29, H₂-12/C-10, H- $12\beta/C-29$, H₂-29/C-10 revealed connection among the partial structures A and C and C-12 via N-11.

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Table 1. ¹H and ¹³C NMR Data of Nakadomarin A (1) in CD₂OD

			0			
position	$\delta_{ ext{H}}{}^{a}$	m	J (Hz)	$\delta_{c}{}^{b}$	m	HMBC ^c (H)
2	-			164.6	s	3, 22, 23
3	5.99	S		105.6	d	$22a^d$
4				137.9	s	3, 6, 8, 9
5				156.2	s	3, 6
6	4.25	brs		75.9	d	12, 13α
7				64.2	s	9 , 12β, 13
8	2.95	dd	2.7, 4.3	43.2	d	9β, 10α, 12α, 13
9α	2.04	dddd	2.4, 4.3, 12.1, 14.2	23.1	t	8, 10
β	1.85	ddd	2.7, 4.5, 14.2			
10 α	2.22	dd	2.4, 12.1	46.6	t	8, 9α, 12, 29
β	2.66	dt	4.5, 12.1			
12 α	3.13	d	12.2	60.6	t	6 , 8 , 10β, 13β,
β	2.39	d	12.2			$29b^e$
13 α	2.17	dd	5.0, 12.7	43.0	t	8, 12, 14, 15 ^e
β	1.65	dd	10.9, 12.7			
14	4.05	m		60.4	d	13β , 15, 16
15	5.61	dd	8.3, 10.1	129.5	d	17α
16	5.97	dt	10.1, 7.2	136.4	d	17α, 1 8 α ^e
17 α	2.42	m		26.4	t	15, 16
β	2.19	m				
18 α	1.76	m		28.9	t	17α, 19 ^e
β	1.51	tt	4.3, 13.0			
19	1.77^{b}	m		25.3	t	17, ^e 18α ^e
20 α	2.93	m		51.7	t	19 ^e
β	3.20	m				
22	2.77	ddd	2.9, 7.4, 14.6	30.0	t	3, ^e 24
	2.76	ddd	3.0, 10.1, 14.6			
23	2.56	m		29.2	t	22, 24, ^e 25
	2.22	m				
24	5.28	ddd	7.0, 8.8, 10.8	129.8	d	22, 23b, ^e 26b
25	5.50	dt	10.8, 7.8	132.7	d	23b, ^e 26, 27a ^e
26	1.97	m		27.6	t	24, 25, 27, ^e 28 ^e
	1.70	m				
27	1.13	m		29.7	t	25, 26, 28, ^e 29
	0.96	m				
28	1.39	m		27.4	t	26, 27, 29b
	1.17	m				
29	2.47	dt	11.9, 3.7	59.6	t	10 α , 12 β , 28 e
	2.36	dt	11.9, 3.6			

^a Recorded at 600 MHz. ^b Recorded at 125 MHz. ^c Delay time (Δ) for C-H long-range coupling was set to 50 ms. ^{*d*} a and b denote upfield and downfield resonances, respectively, of a geminal pair for C-22, C-23, C-26, C-27, C-28, and C-29. ^e These correlations were observed in D-HMBC¹⁰ spectrum ($\Delta = 80$ ms).

The following differential NOE experiments justified linking of C-6, C-14, and C-20 via N-21: irradiation of H-6 (δ 4.25) yielded NOE's for H-20 α (δ 2.93, 1.2%) and H-20 β (δ 3.20, 2.3%), while irradiation of H-20 α afforded 2.8% NOE for H-14 (δ 4.05). The presence of an 8/5-fused azabicyclic system was supported by the ¹³C chemical shifts at C-6 (δ 75.9), C-14 (δ 60.4), and C-20 (δ 51.7) of 1, which were similar to those of the corresponding carbons (C-26; δ 77.6, C-34; δ 57.7, C-28; δ 53.3) of manzamine A^{2,3} in CD₃OD. HMBC correlations for H-3/ C-2, H-22/C-2, H-3/C-4, H-6/C-4, H-8/C-4, H-9/C-4, H-3/ C-5, and H-6/C-5 and a long-range ¹H-¹H coupling for H-6/H-8 indicated that the three sp² quaternary carbons at δ 164.6, 137.9, and 156.2 were assignable to C-2, C-4, and C-5, respectively. The low-field ¹³C chemical shifts at C-2 (δ 164.6) and C-5 (δ 156.2) and the unsaturation degree of 1 indicated the presence of an ether linkage between C-2 and C-5, which was supported by the UV absorption at 228 nm (ϵ 10000).¹¹ Thus the structure of nakadomarin A was elucidated to be 1.

The relative configurations of all chiral centers of nakadomarin A (1) as well as conformation of each ring



≺---+ NOESY

Figure 1. Relative stereochemistry for pentacyclic core of nakadomarin A (1) based on the NOESY correlation and proton couplings. The coupling constants for this moiety (H/H in Hz) are as follows: $8/9\alpha = 4.3$, $8/9\beta = 2.7$, $9\alpha/10\alpha = 2.4$, $9\alpha/10\beta = 12.1$, $9b/10\alpha = <1$, $9\beta/10\beta = 4.5$, $13\alpha/14 = 5.0$, $13\beta/10\beta = 10$, $10\beta/10\beta = 10$ 14 = 10.9, 14/15 = 8.3, and 15/16 = 10.1.

were elucidated on the basis of NOESY data, differential NOE experiments, and ¹H-¹H coupling constants (Figure 1). NOESY cross-peaks for H-9 α /H-12 α and H-10 α /H- 12α and ${}^{1}\text{H}{-}{}^{1}\text{H}$ coupling constants of H-8/H-9 α (J = 4.3Hz), H-8/H-9 β (2.7 Hz), H-9 α /H-10 α (2.4 Hz), H-9 α /H- 10β (12.1 Hz), H-9 β /H-10 α (<1 Hz), and H-9 β /H-10 β (4.5 Hz) showed that the piperidine ring (C-7-C-12) had a boat conformation with H-9 α and H-12 α occupying flagpole orientations.¹² β -Axial orientation of the lone pair at N-11 was deduced from NOESY correlations for H-10 α /H₂-29 and H-12 α /H₂-29. The boat conformation of the piperidine ring elucidated by the NOESY data and ¹H-¹H coupling constants corresponded well to the most stable conformation of a piperidine ring, which was afforded by conformational search¹³ of the 5/5/15/6tetracyclic moiety using MacroModel version 5.014 (Figure 2). The 7,8-cis configuration was indicated by NOESY correlations observed for H-6/H-12 β , H-8/H-13 α , and H-8/ H-14. NOESY data for H-6/H-13 β , H-6/H-15, H-6/H-20 β , H-8/H-14, and H-13 β /H-14 suggested that H-6, H-14, and the lone pair at N-21 were oriented β , α , and α , respectively. Considering the NOE's for H-14/H-17 α , H-14/H-19α, H-14/H-20α, H-16H-17β, H-16/H-18β, and H-16/H-19 β , the eight-membered ring (C-14–N-21) seemed to be an envelope-boat conformation,¹⁵ similar to the N-27 to C-34 ring in manzamine A.^{1a,5} The conformation of the eight-membered ring in 1 deduced from NOE's was close to the most stable conformation of the 8/5-fused azabicyclic system calculated by MacroModel version 5.0 (Figure 3). Thus the relative stereostructure and ring conformation of nakadomarin A were concluded to be 1.

Nakadomarin A (1) is a novel hexacyclic alkaloid consisting of an unprecedented 8/5/5/15/6 ring system. Since nakadomarin A (1) possesses a piperidine ring with

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Figure 2. Most stable conformation (total energy, 47.6 kcal/ mol) of 5/5/15/6-tetracyclic system in nakadomarin A (1) calculated by Macromodel ver. 5.0.



Figure 3. Most stable conformation (total energy, 28.7 kcal/ mol) of 8/5-bicyclic system in nakadomarin A (1) calculated by Macromodel ver. 5.0.

a C₈ alkyl chain attached at N-11 and an 8/5-fused azabicyclic system, which are common structures in manzamine A and ircinal A⁵ nakadomarin A (1) is considered to be a novel class of manzamine-related alkaloid containing a furan ring. A plausible biogenetic path of nakadomarin A (1) through ircinal A is shown in Scheme 1, in which cleavage of the C-2-C-6 bond of ircinal A followed by rotation of the C-4-C-8 bond, and then formation of the C-5-C-6 bond gives a pentacyclic intermediate, which finally yields nakadomarin A (1) after formation of an ether ring and dehydroxylation. Nakadomarin A (1) showed cytotoxicity against murine lymphoma L1210 cells (IC₅₀ 1.3 μ g/mL) and inhibitory activity against cyclin dependent kinase 4 (IC₅₀ 9.9 μ g/ mL). Compound 1 exhibited antimicrobial activity against a fungus (Trichophyton mentagrophytes, MIC 23 µg/mL) and a Gram-positive bacterium (Corynebacterium xerosis, MIC 11 μg/mL).

Experimental Section¹⁶

Sponge Materials. The medium brown color sponge *Amphimedon* sp. (order Haplosclerida; family Niphatidae) was collected off Kerama Islands, Okinawa, and kept frozen until





used. The piece of sponge is a small mound with irregular meandering ridged surface. Texture is firm and compressible. Mesohyl consists of a fibrous reticulation with more dense plumoreticulate fiber centrally. Spicules possess few interstitial. Spicule fans extend slightly beyond the surface at right angles. Primary fibers possessing 80-100 mm thick are centrally cored by 5-12 spicules. Secondary fibers (80-100 mm) are cored by 1-3 spicules. Megascleres are small oxeas (mean size; 187×9 mm), which are generally straight or slightly curved. There is no microsclere. The voucher specimen (SS-264) was deposited at the Faculty of Pharmaceutical Sciences, Hokkaido University.

Extraction and Isolation. The sponge (1.0 kg, wet weight) was extracted with MeOH (1 L \times 2). The methanolic extract (71 g) was partitioned between ethyl acetate (400 mL \times 3) and 1 N NaCl aq. Part (15 g) of the EtOAc soluble material (51.1 g) was subjected to SiO₂ columns three times (solvent system; CHCl₃/MeOH, 90:10, cyclohexane/acetone/Et₂-NH, 90:10:2, CHCl₃/MeOH, 90:10) to yield nakadomarin A (1, 6.0 mg, 1.8 \times 10⁻³ % wet weight).

Nakadomarin A (1): a colorless amorphous solid; $[\alpha]^{25}_{\rm D}$ -16° (*c* 0.12, MeOH); UV (MeOH) $\lambda_{\rm max}$ 206 (ϵ 11000) and 228 nm (10000); IR (KBr) $\nu_{\rm max}$ 2920, 2850, 1455, and 1080 cm⁻¹; ¹H and ¹³C NMR (see Table 1); EIMS *m*/*z* 392 (M⁺) and 296; HREIMS *m*/*z* 392.2826 (M⁺), calcd for C₂₆H₃₆N₂O, 392.2828.

Computational Methods. Conformational searching was carried out using Pseudo Monte Carlo simulation in Macromodel program. The closure bonds for the 5/5/15/6-tetracyclic system were chosen at C-9–C-10 and C-27–C-28 with the closure limit from 1 to 4 Å, while the closure bond for 8/5-bicyclic system was chosen at C-18–C-19. Five thousand Monte Carlo steps were performed and produced 19 and 2 conformers for 5/5/15/6-tetracylcic and 8/5-bicyclic systems, respectively, which were obtained within 3 kcal/mol of the lowest energy conformers. Each conformer was finally minimized by molecular mechanics calculation of MM2* force field in H₂O.

⁽¹⁶⁾ Analytical instruments and general procedures in this work were described in the previous report: Kobayashi, J.; Tsuda, M.; Fuse, H.; Sasaki, T.; Mikami, Y. *J. Nat. Prod.* **1997**, *60*, 150–154.

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Supporting Information Available: NMR spectra and PDB files of **1** (11 pages). This material is contained in libraries on microfiche, immediately follows this article in the microfilm version of the journal, and can be ordered from the ACS; see any current masthead page for ordering information. JO9715377